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Photoluminescence and radiation effect of Er and Pr implanted silicon-rich silicon oxide thin films

Fang Zhu^a, Zhisong Xiao^{a,*}, Lu Yan^a, Feng Zhang^a, Kun Zhong^b, Guoan Cheng^b

^a Department of Physics, School of Science, Beihang University, Beijing 100191, China
^b Department of Materials Science and Engineering, Beijing Normal University, Beijing 100875, China

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ABSTRACT

Er and Pr ions were implanted into silicon-rich silicon oxide (SRSO) thin films with Si crystals embedded in SiO₂ matrix. The 525 and 546 nm luminescence peaks were clearly observed in Er-only doped film, but disappeared in the photoluminescence (PL) spectra of Er–Pr codoped films. Instead, a broad PL spectrum extending from 450 to 700 nm was obtained for Er–Pr codoped films with Er/Pr concentration ratio of 1. Concentration profiles of Si, Er and Pr ions in films were simulated by SRIM2006 and related radiation effect on PL response was also discussed. Our results indicate that this material is a potential candidate for the development of white light-emitting diode (LED) and field emission displays for its visible luminescence.

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BEAM INTERACTIONS WITH MATERIALS AND ATOMS

1. Introduction

Considerable research has been focused on the visible light emitting using rare earth (RE) ions such as Er^{3+} , Pr^{3+} , Eu^{3+} , Nd^{3+} , Ho^{3+} and Tm^{3+} doping in different hosts in the past decades [1– 4]. White light source with high emissivity is required for display and lighting technology, which can be generated either by precise mixing of primary colors, i.e. blue, green, and red colors or by light with complementary colors. Efforts to simulate the white light emission by controlling the intensities of different red–green–blue (RGB) lights have been reported by researchers using RE ions doped in different lattices. Silva et al. obtained white light in Tm/ Er/Yb codoped fluoride glasses [5]. Recently Gouveia et al. demonstrated white emission in Tm/Ho/Yb codoped fluorolead germinate glass with high color purity (blue: 97%; green: 100%; red: 95%) [6].

Among those RE ions, Er^{3+} is the most popular one due to its large efficiency in the region of prime colors. Pr^{3+} has also been found much attractive as it offers the efficient IR emission for optical amplification at 1.3 µm and RGB emission simultaneously [7,8]. To improve the luminescence efficiency, RE ions have been implanted in silicon-rich silicon oxide (SRSO) thin films. A particular advantage of RE ions doped in SRSO thin films is that the excitation of RE ions occurs predominantly through an Auger-type interaction between carriers in nc-Si and the RE ions [9]. Si nano-crystalline (nc-Si) were demonstrated to act as photosensitizers for molecular oxygen and several kinds of rare earth ions. In particular, erbium ions can be excited very efficiently by the energy transfer from nc-Si. The effective absorption cross-section of the intra-4*f* shell transition of Er^{3+} was enhanced by 2–4 orders of magnitude because of the large absorption cross-section of nc-Si in visible range and high efficient energy transfer from nc-Si to Er^{3+} [10].

In the present work we have demonstrated the generation of violet–green–red emission from a combination of Er–Pr codoped in SRSO thin film. Concentration profiles of Si, Er and Pr ions in films were simulated by the SRIM2006 and related radiation effect on PL response was also discussed.

2. Experiment

Wet oxygen oxidation was performed to produce SiO₂ film on Si(1 0 0) wafers at 1000 °C for 1.5 h. The thickness of resultant SiO₂ film was more then 1 µm, which was calculated by silicon oxidation dynamics formulation. SiO₂ films were implanted in vacuum with 3×10^{17} cm⁻² Si ions at 45 keV by using a metal vapor vacuum arc (MAVVA) implanter. Then 2×10^{16} cm⁻² Er ions were implanted at 60 keV. Finally, Pr ions were implanted with ion dose of 2, 4, 8, 10 and 20×10^{16} cm⁻² at 60 keV. The resulting Pr peak concentration was varied from 2×10^{20} to 2×10^{21} cm⁻³. The ion beams were implanted with an incident angle of ~10° with respect to the substrate in order to minimize ion-channeling effects. The implanted samples were then annealed in air for 2 h at 800 °C.

The annealed sample implanted with only Si ions was analyzed by X-ray diffraction (XRD) using a $\theta - 2\theta$ diffractometer excited with a Cu K α I (λ = 0.15406 nm) radiation as well as a step size of 0.02°. Photoluminescence (PL) of the samples was measured using a fluorescence photospectrometer at room temperature. The PL signal was detected by CCD detector. All spectra were excited at the wavelength of 325 nm.

3. Results and discussion

Si ions were implanted into α -SiO₂ thin films followed by 800 °C annealing. Crystalline (1 1 1) Si (c-Si) was formed as shown in XRD spectrum (Fig. 1). Nano-crystalline Si (nc-Si) is also possibly existed in those films as reported by Xiao et al. [11]. The introduced c-Si provided the system with a large absorption cross-section $(10^{-16}-10^{-17} \text{ cm}^{-2})$ and acted as sensitizers for the RE ions. This excitation process has been modeled as a dipole-dipole interaction between the c-Si (nc-Si) and Er³⁺ with characteristic energy transfer times less than 1 µs [12]. Kik et al. reported that the maximum activated Er³⁺ concentration by Si nanocrystals was of about $1 \times 10^{19} \, \text{cm}^{-3}$ for the samples annealed at high temperature (1100 °C). They attributed such saturation to Er clustering effect or to Auger deexcitation [13]. More recently, Wojdak et al. found that in highly doped samples ([Er] = 2×10^{20} cm⁻³) the activated Er³⁺ fraction was less than 3%. Franzò et al. observed an enhanced Er³⁺ emission by reducing processing temperature of Er-doped SRSO to 800 °C. In this case there are amorphous Si nanoclusters in the processed layers, which appear to act as better Er-sensitizing centers than the crystalline Si by showing a larger density and a concomitant reduction of the relative distance between Er ions and nc-Si [14].

Fig. 2 shows the luminescence of Er-only doped SRSO film. The green emission at 525 and 546 nm is originated from the $\text{Er}^{3+2}\text{H}_{11/2} \rightarrow {}^{4}\text{I}_{15/2}$ and ${}^{4}\text{S}_{3/2} \rightarrow {}^{4}\text{I}_{15/2}$ transition, respectively. The red PL peak at 662 nm is due to the ${}^{4}\text{F}_{9/2} \rightarrow {}^{4}\text{I}_{15/2}$ transition of Er^{3+} . There are also 380 nm violet PL peak and near infrared 769 nm emission due to ${}^{2}\text{H}_{9/2} \rightarrow {}^{4}\text{I}_{15/2}$ transition, respectively. The broad emission band ranging from 450 to 700 nm is proposed to be the contribution of amorphous and c-Si (nc-Si) luminescence in SiO₂ matrix. The peak of this broad PL band may shift from sample to sample, depending on the band gap (related to the size) of nc-Si. Liu et al. observed the peak shift from 590 nm (for a 700 °C annealed sample) to 750 nm (for an 1100 °C annealed sample) [15].

Fig. 3 shows the PL spectra of Er–Pr codoped SRSO films. Comparing with the PL spectra of Er-only doped film, the broad lumi-



Fig. 2. PL spectra of Er-only doped SRSO film.

nescence band of 450–700 nm was clear for the film with Pr/Er concentration ratio of 1 (Pr ion dose of 2×10^{16} cm⁻²). A new green emission peak at 568 nm is originated from Pr³⁺ ³P₁ \rightarrow ³H₅ transition and the red emission of 618 nm is due to Pr³⁺ ³P₁ \rightarrow ³H₆ transition. The shoulder PL peak is locating at ~500 nm is suggested to be the transition of Pr³⁺ from ³I₀ + ³P₁ state to ³H₄ or ³H₅ state. By Pr codoping the Er³⁺ emission at 525, 546 and 662 nm disappeared or overlapped by the broad luminescence band of 450–700 nm. Instead, Pr³⁺ emission at 495, 568 and 618 nm were observed. It could be understood by the energy transfer process from Er³⁺ to Pr³⁺ as following.

After absorption of the pumping photons Er^{3+} rapidly relaxed into ${}^{2}H_{11/2}$, ${}^{4}S_{3/2}$ and ${}^{4}F_{9/2}$ state through nonradiative processes, which have been indicated in [5]. The Pr ions after absorption of the pumping photons could relax to ${}^{3}F_{4}$ state by a nonradiative transition, and then relaxed the ground state, corresponding to the well known 1.3 µm emission [7]. The $Er^{3+} {}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$ transition is nearly resonant with the $Pr^{3+} {}^{3}F_{4} \rightarrow {}^{3}P_{2}$ transition with an energy mismatch less than 0.1 eV. The energy levels involved for observed PL bands of Er^{3+} ions decayed nonradiatively from the ${}^{3}P_{2}$ state back to the ${}^{3}P_{1}$ nearly instantaneously, since the multiphonon effects in silica would result in a nonradiative $Pr^{3+} {}^{3}P_{2} \rightarrow {}^{3}P_{1}$ transition at a much fast rate, which enhanced the emission of Pr^{3+} 568 and 618 nm. We simulated concentration profiles of Si, Er and Pr ions and introduced vacancies in films by SRIM2006. Fig. 5 shows that Er ions and Pr ions located in the same range



Fig. 1. XRD spectrum of SiO₂ sample implanted with only Si.



Fig. 3. PL spectra of Er-Pr codoped SRSO film with [Er]/[Pr] concentration ratio of 1.



Fig. 4. Schematic energy level diagram of Er³⁺ and Pr³⁺ showing different transitions in Er–Pr codoped SRSO films.

80 ± 50 nm under surface. Since Pr^{3+} and Er^{3+} distribution profiles overlapped, it is easy for the energy transfer occurring between nearby Pr^{3+} and Er^{3+} .

The broad emission band ranging from 450 to 700 nm was not affected by the presence of Pr^{3+} , which means this PL band was originated from amorphous and c-Si (nc-Si) luminescence in SiO₂ matrix. Since Er–Pr codoped samples high dose RE ions were implanted in SRSO film ([Er] = 2×10^{20} cm⁻³, [Pr] = 2×10^{20} cm⁻³), above PL performance of Pr³⁺ and Er³⁺ can be understood as following. High concentration c-Si (nc-Si) provided a large absorption cross-section and acted as sensitizers for the RE ions. It contributed to an efficient route for efficient energy transfer of Pr³⁺ and Er³⁺.

In Fig. 3, when the Pr³⁺ ion dose was 4×10^{16} cm⁻² with Er/Pr concentration ratio of 1/2, corresponding to Pr³⁺ concentration in films up to 4×10^{20} cm⁻³, the PL intensity for RE ions decreased. There is only a week peak at 495 nm corresponding to Pr³⁺ ${}^{3}I_{0} + {}^{3}P_{1} \rightarrow {}^{3}H_{4}$ transition. It has been demonstrated the concentration quenching effect happened when high dose of Pr ions were implanted. On the other hand, the PL from RE ions disappeared because of the clustering effect, which decreased the 4*f* shell emission from trivalent RE ions. The broad emission band ranging



Fig. 5. Concentration profiles of various ions in films simulated by the SRIM2006.

from 450 to 700 nm originated from amorphous and c-Si (nc-Si) in SiO₂ matrix disappeared. It is proposed that great sputtering effect peeled off the c-Si (nc-Si) in the surface through high dose of Er–Pr ion implantation, and the RE metal layer was formed in the surface of the films. It thus prevented the light emission from c-Si (nc-Si) in SiO₂ matrix.

4. Conclusions

Er and Pr codoped silicon-rich silicon oxide films were prepared. A broad PL spectrum extending from 450 to 700 nm was obtained for Er–Pr codoped samples with Er/Pr concentration ratio of 1 pumped at 325 nm. Er³⁺ was demonstrated to act as sensitizer for Pr³⁺ in the codoped films, and the green–red band luminescence has been enhanced by energy transfers from Er³⁺ to Pr³⁺. The broad emission band ranging from 450 to 700 nm is proposed to be the contribution of amorphous and c-Si (nc-Si) in SiO₂ matrix. A metal layer can be formed by the radiation progress which can deteriorate the luminescence. It thus implies that Pr–Er codoped SRSO films is a promising material for application in white light-emitting diode (LED) and field emission displays.

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